Page 5

AMENDMENTS TO THE CLAIMS:

This listing of claims will replace all prior versions, and listings, of claims in the

application:

LISTING OF CLAIMS:

1. (Original) A mesostructured material comprising a mineral phase within which are

dispersed particles of nanometric dimensions comprising at least one metal oxide in the

crystalline state selected from a cerium oxide, a zirconium oxide, a titanium oxide and an

oxide of a rare earth other than cerium, said oxide comprising at least one metallic element

M in the cationic form, in solid solution within the crystalline lattice of said oxide.

2. (Currently Amended) A material according to claim 1, characterized in that it

which is thermally stable.

3. (Currently Amended) A material according to claim 1 or claim-2, characterized in

that wherein at least at a local level, it has one or more mesostructures selected from

mesoporous mesostructures with three-dimensional hexagonal P63/mmc symmetry, with

two-dimensional hexagonal symmetry, with three-dimensional cubic la3d, Im3m or Pn3m

symmetry; from vesicular or lamellar type mesostructures, or from vermicular type

mesostructures.

4. (Currently Amended) A material according to any one of claims 1 to 3 claim 1,

characterized in that wherein said particles with nanometric dimensions are particles with a

spherical or isotropic morphology at least 50% of the population of which has a mean

diameter in the range 1 to 10 nm, or highly anisotropic rod type particles at least 50% of the

Page 6

population of which has a mean transverse diameter in the range 1 to 10 nm and a mean

5. (Currently Amended) A material according to any one of claims 1 to 4 claim 1,

characterized in that wherein the metal oxide present within said particles with nanometric

dimensions has a degree of crystallinity of 30% to 100% by volume.

6. (Currently Amended) A material according to any one of claims 1 to 5 claim 1,

characterized in that <u>wherein</u> the quantity of cations of element M in solid solution (or, if

appropriate, of the totality of the solid solution doping agents) represents at least 0.2% of the

total quantity of metallic cations present in the oxide.

length that does not exceed 100 nm.

7. (Currently Amended) A material according to any one of claims 1 to 6 claim 1,

characterized in that <u>wherein</u> said particles with nanometric dimensions are particles based

on cerium oxide, and in that said element M is selected from rare earths other than cerium,

transition metals that are capable of being integrated in the cationic form in solid solution into

a cerium oxide, and alkaline-earth metals.

8. (Currently Amended) A material according to any one of claims 1 to 6 claim 1,

characterized in that <u>wherein</u> said particles with nanometric dimensions are particles based

on zirconium oxide, and in that said element M is selected from rare earths, transition metals

that are capable of being integrated in the cationic form in solid solution into a zirconium

oxide, and alkaline-earth metals.

9. (Currently Amended) A material according to any one of claims 1 to 6 claim 1,

characterized in that wherein said particles with nanometric dimensions are particles based

on titanium oxide, and in that said element M is selected from rare earths, transition metals

<u>assigned</u>

Page 7

that are capable of being integrated in the cationic form in solid solution into a titanium oxide,

and alkaline-earth metals.

10. (Currently Amended) A material according to any one of claims 1 to 6 claim 1,

characterized in that wherein said particles with nanometric dimensions are particles based

on an oxide of a rare earth other than cerium, and in that said element M is selected from

rare earths other than the rare earth constituting said oxide, transition metals that are

capable of being integrated in the cationic form in solid solution into a rare earth oxide, and

alkaline-earth metals.

11. (Currently Amended) A material according to any one of claims 1 to 10 claim 1,

characterized in that wherein said mineral phase is at least partially constituted by silica.

12. (Currently Amended) A material according to any one of claims 1 to 11 claim 1,

characterized in that wherein the mineral phase also comprises metallic cations of metal M

ad/or clusters based on metal M dispersed within said mineral phase and/or on the surface

of said mineral phase.

13. (Currently Amended) A material according to any one of claims 1 to 12 claim 1,

characterized in that <u>wherein</u> at least a portion of the particles with nanometric dimensions

dispersed within the mineral binder phase is in contact with porous portions constituting the

internal space of the material.

14. (Currently Amended) A material according to any one of claims 1 to 12 claim 1,

characterized in that wherein the (mineral binder phase/particles with nanometric

dimensions) molar ratio is in the range 20:80 to 99.5: 0.5.

15. (Currently Amended) A material according to any one of claims 1 to 14 claim 1; characterized in that it which comprises crystallites based on the oxide, hydroxide, oxyhydroxide, carbonate or hydroxycarbonate of said element M.

16. (Currently Amended) An ordered mesoporous or mesostructured material according to any one of claims 1 to 15 claim 1, characterized in that wherein said material has a BET specific surface area in the range 750 to 2300 m² per cm³ of material.

17. (Currently Amended) A process for preparing a material according to any one of claims 1 to 16 claim 1, characterized in that it which comprises successive steps consisting in comprising:

- a) producing a mineral mesostructure integrating, within its walls, particles with nanometric dimensions comprising a metal oxide in its crystalline state selected from a cerium oxide, a zirconium oxide, a titanium oxide and a rare earth oxide other than cerium;
- b) introducing into the mesoporous structure obtained, a compound based on said element M, the total amount of element M introduced into the structure with respect to the total surface area developed by the mesostructure being less than 5 micromoles of cation per m² of surface; and
- c) subjecting the mesostructure produced to a temperature of at least 300°C and not higher than 1000°C.
- 18. (Currently Amended) A preparation process according to claim 17, characterized in that which step a) is implemented by carrying out the following steps:
- a1) forming an initial medium comprising a templating agent, namely a surfactant type amphiphilic compound which can form micelles in the reaction medium;

Page 9

a2) adding to the medium of step 1a) a colloidal dispersion of particles with nanometric dimensions based on a metal oxide in the crystalline state, selected from cerium

oxide, a zirconium oxide, a titanium oxide and a rare earth oxide other than cerium;

a3) forming a mesostructured mineral phase, usually at least partially, or even essentially

constituted by silica, said mineral phase by adding a mineral precursor to the

medium; and

eliminating the templating agent, in particular by heat treatment or by entrainment by

a solvent.

a4)

19. (Currently Amended) A preparation process according to claim 17 or claim 18,

characterized in that wherien step b) is carried out by immersing the mesostructured material

obtained at the end of step a) in a solution comprising the element M in a concentration in

the range 0.1 to 1.5 mol.1 then filtering the medium obtained.

20. (Currently Amended) A preparation process according to claim 17 or claim 18,

characterized in that wherein step b) is carried out by immersing the mesostructured material

obtained at the end of step a) in an aqueous or hydro-alcoholic solution comprising cations

of metal M in a concentration in the range 0.2 to 1.5 mol/1 then centrifuging the medium

obtained at a rate of 2000 to 5000 rpm, for a period not exceeding 30 minutes.

21. (Currently Amended) A preparation process according to any one of claims 17 to

20 claim 17, characterized in that wherein, following the impregnation/heat treatment

procedures of steps b) and c), it comprises one or more subsequent impregnation/heat

treatment cycles implementing steps of type b) and c) carried out on the solid obtained from

the preceding cycle.

Dece 40

Page 10

22. (Currently Amended) Use of a material according to any one of claims 1 to 16 or

of a A material that can be obtained by the process of any one of claims 17 to 21, as claim

17, which is a heterogeneous acidic, basic or redox catalyst.

23. (Currently Amended) Use of a material according to any one of claims 1 to 16 or

of a material that can be obtained by a process according to any one of claims 17 to 21, in

which said A material comprises comprising particles of cerium oxide integrating manganese

in solid solution within the walls of its mesostructure, as a catalyst for absorption of oxides of

nitrogen.

24. (Currently Amended) Use of a material according to any one of claims 1 to 16 or

of a A material that can be obtained by a process according to any one of claims 17 to 21

<u>claim 1</u>, as a support for catalytic species.

25. (Currently Amended) A catalyst that can be obtained by supporting catalytic

species on a material according to any one of claims 1 to 16 or of a material that can be

obtained by a process according to any one of claims 17 to 21 claim 1.